

ELECTROMAGNETIC FUNCTIONALIZED POLYPYRROLE NANOCOMPOSITES CONTAINING MAGNETIC NANOFILLERS OF HIGH ASPECT RATIOS

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This work reports electromagnetic functionalized PPy nanocomposites containing magnetic nanofillers with high aspect ratios such as α -FeOOH nanorods and γ -Fe₂O₃ nano-needles. The morphologies and structures of the PPy nanocomposites were characterized by transmission electronic microscope (TEM) and X-ray diffraction (XRD). The room-temperature conductivity of the nanocomposites was studied by a four-probe method. And the magnetic properties were investigated at room temperature using a vibrating sample magnetometer with a maximum magnetic field of 10 kOe.

Introduction

Polypyrrole (PPy) nanocomposites have attracted particular attention because of their high conductivity, redox properties, and environment stability.^{1,2} Recently, electromagnetic functionalized PPy nanocomposites have attracted great interests owing to their promising applications in gas and humidity sensors,^{4,5} electromagnetic interference shielding,⁶ electrochemical storage,^{7,8} lithium insertion electrode⁹ and so on. Therefore, preparation and study of electromagnetic functionalized PPy nanocomposites is of great significance.

In this work, the preparation of electromagnetic functionalized PPy nanocomposites containing magnetic nanofillers with high aspect ratios such as α -FeOOH nanorods, and γ -Fe₂O₃ nano-needles is reported. Then, the morphologies, structures, electrical

conductivity and magnetic properties are studied for the PPy nanocomposites.

Experimental

Pyrrole (Py) was distilled and stored under N₂ gas. Other reagents including NaOH, FeCl₃·6H₂O, FeSO₄·7H₂O, p-TSA dopant and γ -Fe₂O₃ nano-needles were used without further purification. The γ -Fe₂O₃/PPy nanocomposites were synthesized by a template-free method associated with ferromagnetic γ -Fe₂O₃ needles as the hard-template.¹⁰ Typical synthesis process of α -FeOOH/PPy nanocomposites is as follows:¹¹ 1 ml of pyrrole monomer and a given amount of FeSO₄·7H₂O were mixed with the given amount of NaOH dissolved in 40 mL of de-ionized water under mechanical stirring at room temperature. The mixture was then cooled down to 0~5°C in an ice bath. After 20 minutes, 30 mL of FeCl₃·6H₂O solution (1 mol/L) was added drop by drop to the above mixture and the polymerization took place under stirring for 4 hours. The final product was washed respectively with de-ionized water and ethanol for several times. Finally, the product was dried at 50°C under vacuum for 24 h.

X-ray diffraction was carried out on a RINT 2000 Wilder-angle goniometer. The morphology was characterized by a transmission electron microscope (TEM, Hitachi-9000). The room-temperature conductivity for pressed products was measured by a standard four-probe method with a Keithley 196 SYSTEM DM digital multi-meter and

ADVANTEST R6142 programmable dc voltage/current source. A vibrating sample magnetometer (VSM, Lakeshore 7307, USA) was employed to measure the magnetic properties of the products.

Results and discussion

TEM images as shown in Fig. 1a display that the PPy is coated on the γ -Fe₂O₃ nano-needles (a diameter of about 30 nm). The γ -Fe₂O₃/PPy is a kind of core-shell nanocomposites. TEM images as shown in Fig. 1b display that the α -FeOOH/PPy nanocomposites contain α -FeOOH nanorods (a diameter of about 3-5 nm and a length of about 20-40 nm) and PPy nanospheres (a diameter of about 140 nm).

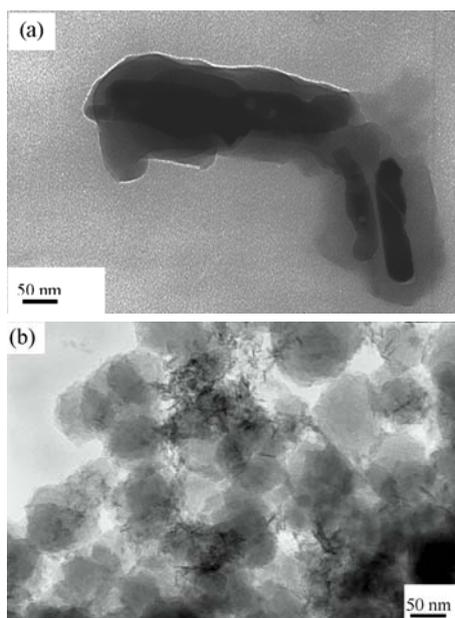


Fig. 1 TEM images of (a) γ -Fe₂O₃/PPy and (b) α -FeOOH/PPy nanocomposites.

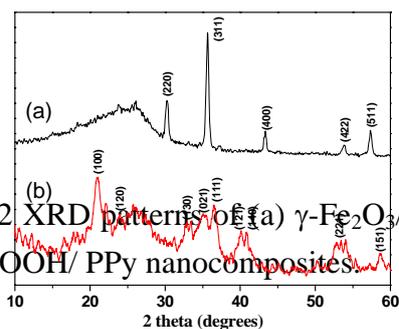


Fig. 2 XRD patterns of (a) γ -Fe₂O₃/PPy and (b) α -FeOOH/PPy nanocomposites.

Except a broad peak centered at $2\theta = 25.26^\circ$ which is the characteristic peak of PPy as shown in Fig. 2a,b, the sharp peaks for Fig. 2a and Fig. 2b can be well indexed to the γ -Fe₂O₃ (JCPDS 39-1346) and α -FeOOH (JCPDS 29-713), respectively.

The conductivity of the γ -Fe₂O₃/PPy and α -FeOOH/PPy is 50 and 2.85 S/cm, respectively.

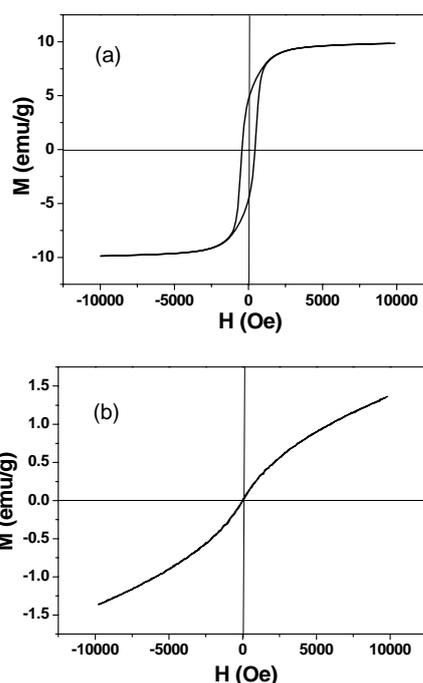


Fig. 3 Hysteresis loops of (a) γ -Fe₂O₃/PPy and (b) α -FeOOH/PPy nanocomposites.

Fig. 3 shows the dependence of the magnetism (M) on the applied magnetic field (H) for (a) γ -Fe₂O₃/PPy and (b) α -FeOOH/PPy nanocomposites. The γ -Fe₂O₃/PPy nanocomposite exhibits excellent ferromagnetic behaviors, in which the saturated magnetization (M_s) and coercive force (H_c) are calculated to be 9.86 emu/g and 440.11 Oe, respectively (Fig. 3a). The α -FeOOH/PPy nanocomposite displays antiferromagnetic behaviors. The magnetization at 10 KOe is 1.36 emu/g (Fig. 3b).

Conclusions

The γ -Fe₂O₃/PPy and α -FeOOH/ PPy nanocomposites have been successfully synthesized by a chemical method. The TEM images show that the γ -Fe₂O₃/PPy nanocomposite has a core-shell structure and the α -FeOOH/ PPy nanocomposite is consisted of γ -Fe₂O₃ nanorods and PPy nanospheres. The two kinds of PPy nanocomposites have good electrical and magnetic properties.

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