

CHANGE OF TEXTILE WETTABILITY BY TiO₂ PARTICLES

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Abstract

Surface roughness can change the wetting characteristics of solids. In this study is used SOL-GEL method based on hybrid polymers to stabilize TiO₂ nano particles on fiber surfaces. The wettability of structures was estimated by contact angle measurement. This experimental method was used with different TiO₂ particles.

Key words: SOL-GEL, CONTACT ANGLE, HYDROPHOBIC TREATMENT

1. Introduction

The purpose of the contact angle measurement is to study the surface wettability of the textiles materials as it was on the objectives in this work. As it is known from theory that some fibers are more hydrophilic than others, for examples cotton substrates, had the contact angle for original cotton was 0°, because water drops spread instantly when placed on the surface of the substrate. This is due to the cellulose hydroxyl groups of cotton that make cotton super-hydrophilic. But after treatment with sol-TiO₂ and application hydrophobic finish the cotton substrate showed an increasing of contact angle of above 90° as the concentration of the TiO₂ increases. As it known in theory that a surface with a contact angle of 0° to 30° has hydrophilic property whereas a surface with contact angle 90° and more is super-hydrophobic. The application of sol-TiO₂ finish and hydrophobic finish with TiO₂ nanoparticles on the textile material surface show an improvement in surface properties of the textiles making the surface more rough because of the layer on the surface of the material. This improves the wettability properties of the textile material, and thus increases the contact angle of the material. This are the finding observed in this research work, as it is observed from the graphs obtained for comparing the contact angle with this TiO₂ concentration of different textile materials. The overall observations for the contact angle measurement was that TiO₂ nanoparticles have an effect on the surface textile material as applied finishing methods, and that its effect increases the surface wettability of the textile material. The degree of increase depends on the important property of materials that is controlled by the chemical composition and the geometry of the surface.

The result of a contact angle between liquid, air and a solid state surface with the structure of a solid state surface is given by the contact of the liquid with a jagged surface. [1]

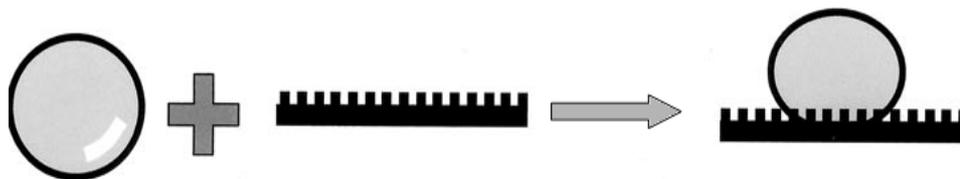


Figure 1. Behavior of a drop of liquid on a jagged surface [1]

2. Experimental part

The main goals of the experimental part was to observe the hydrophobic effect of different TiO₂ powders on textile material when using different treatment methods such as Sol-gel method and the hydrophobic treatment. The TiO₂ powder utilized during the experiment were labeled according to their trade names [091009/1, 181109/2].

Materials used for the experiment were

- glass microfiber
Areal density was 75g.m⁻². Average diameter of fibres is about 1 µm.
- Cotton (30x12 ČSN 80 0101)

2.1 Sol description

Preparation of sol based TMSPM the basis for the preparation of sol was TMSPM ((3 - trimethoxysilylpropylmethacrylate)

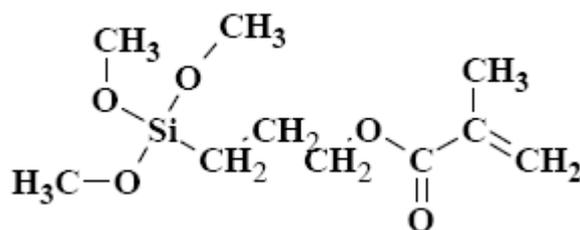


Figure 2. Chemical formula TMSPM

The principle consisted of sol dissolution TMSPM Mid required amount of IPA and the dissolution of other ingredients (water, HCl, BPO) in the second half of the required amount of IPA. Then the two solutions were mixed under vigorous stirring. Resulting sol was heated in boiling under reflux for 30 minutes after which the sol was cooled. Part of the final sol was diluted with IPA in the ratio 1:4. Sol was identified as AC4.

The application conditions of the sol gel with the TiO₂ powder on 100% cotton fabric

- sol gel solution 20ml;
- TiO₂ weights 0.01g, 0.02g, 0.05g and 0.1g;
- Time of sample exposition to the sol gel/TiO₂ powder dispersion 1minute;
- drying at room temperature 20°C for 5 minutes
- The samples were then placed in a 85°C dryer for fixation process for 3 hours

2.3 Hydrophobic Treatment

The hydrophobic solution was prepared for the hydrophobic treatment of textiles. The application was done on 2 different fibres (Cotton, Glass).

Chemicals used for the preparation of the hydrophobic treatment solution

- 30g/l Lukofix T40D
- 20g/l Katalyzator C48
- 10ml/l Acetic acid (CH₃COOH)

Experimental procedure:

- About 30ml of the prepared hydrophobic treatment solution was mixed with different TiO₂ powders at various concentration : 0.1g/l, 0.5g/l, 1g/l, 5g/l, 10g/l, 20g/l
- The mixing of TiO₂ powder was done by the Ultrasound machine under the following settings; AM-30%, time- 2minutes, Energy -3.570KJ

3. Results and discussion

3.1 Contact angle measurement results - the Sol Gel method results

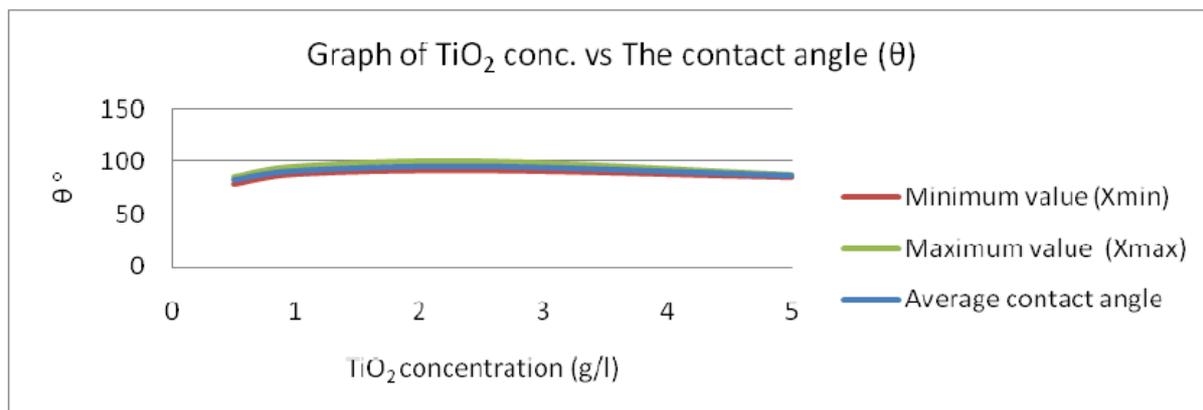


Figure 3. The graph showing the relation between the contact angle with sol-TiO₂ treated cotton surface for TiO₂-091009/1

The surface wettability was assessed by contact measured for sol-TiO₂ cotton treatment, showing the relationship between the sol-TiO₂ concentrations with the contact angle. In figure 3 the 0 concentration of the sol-TiO₂ on the cotton sample was not measurable due to the complete absorption of water by the cotton substrate. The graph distribution above shows the contact angle increasing at the low sol-TiO₂ concentration and decreasing towards the higher concentration.

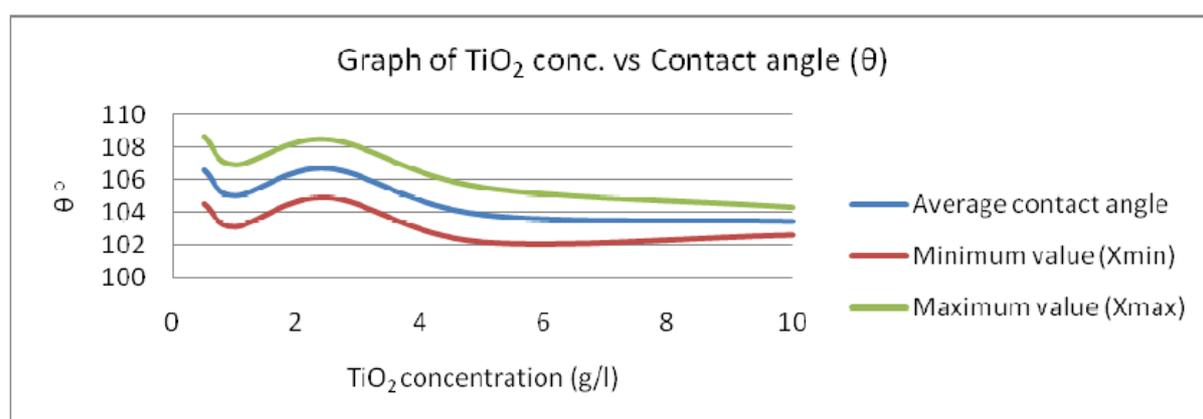


Figure 4. The graph showing the relation between the contact angle with sol-TiO₂ treated cotton surface for TiO₂-181109/2

With the use of TiO₂-181109/2 powder the contact angle distribution with the sol-TiO₂ concentration started with a decrease at 1g/l concentration, then an increase at 2.5g/l concentration and further decline to a constant distribution till high concentration.

3.2 Hydrophobic Treatment Results

Hydrophobic Treatment of Cotton with different TiO₂ powders

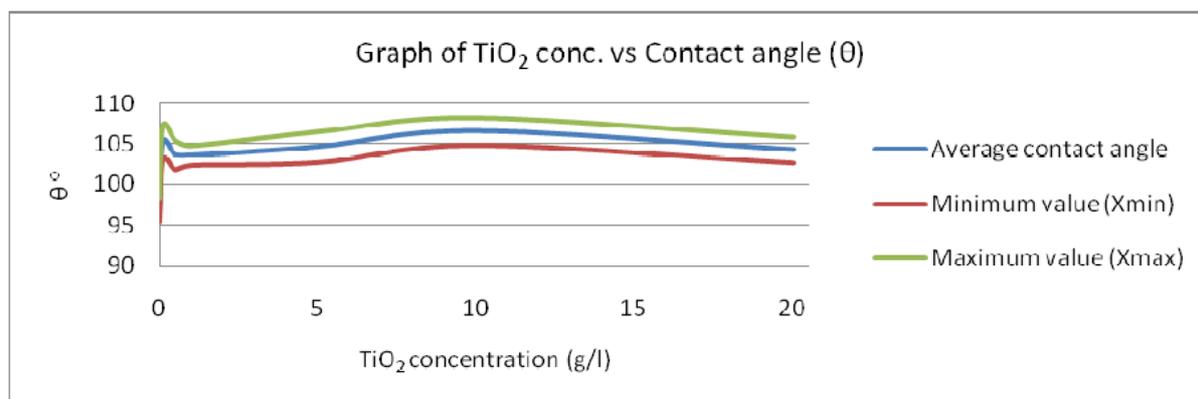


Figure 5. The graph of contact angle with hydrophobic-TiO₂ treated cotton surface for TiO₂-091009/1

The figure above shows the graph distribution of the contact angle, in which the wettability of the cotton surface is assessed on the hydrophobic-TiO₂ treated cotton surface. The TiO₂-091009/1 powder in hydrophobic treatment shows the contact angle increasing from the low TiO₂ concentration and to the maximum increase at 10g/l, then a decrease as the concentration decreases.

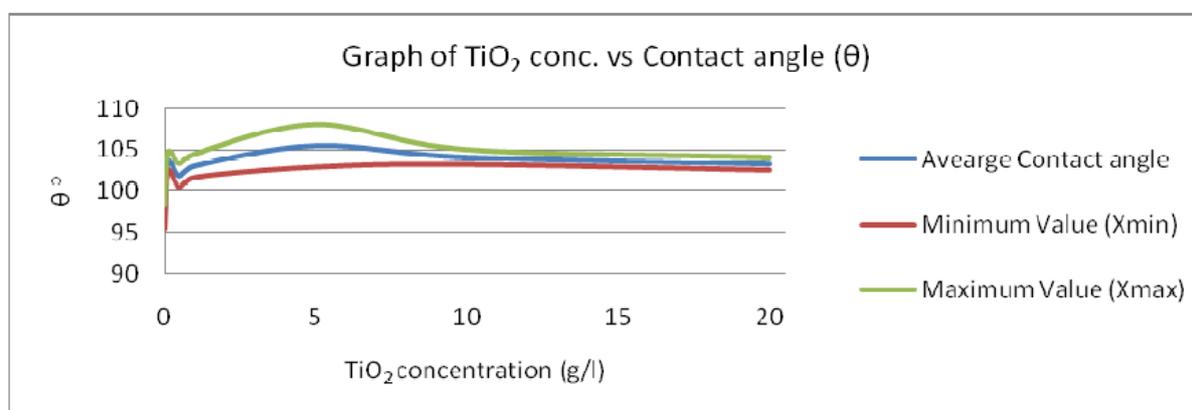


Figure 6. The graph of contact angle with hydrophobic-TiO₂ treated cotton surface for TiO₂-81109/2

For the distribution in fig.6 the contact angle is showing its maximum distribution at concentration of 5g/l and the decrease to a constant distribution as the concentration increases. That is the contact angle is more or less the same for the concentration of TiO₂ from 10g/l to 20g/l.

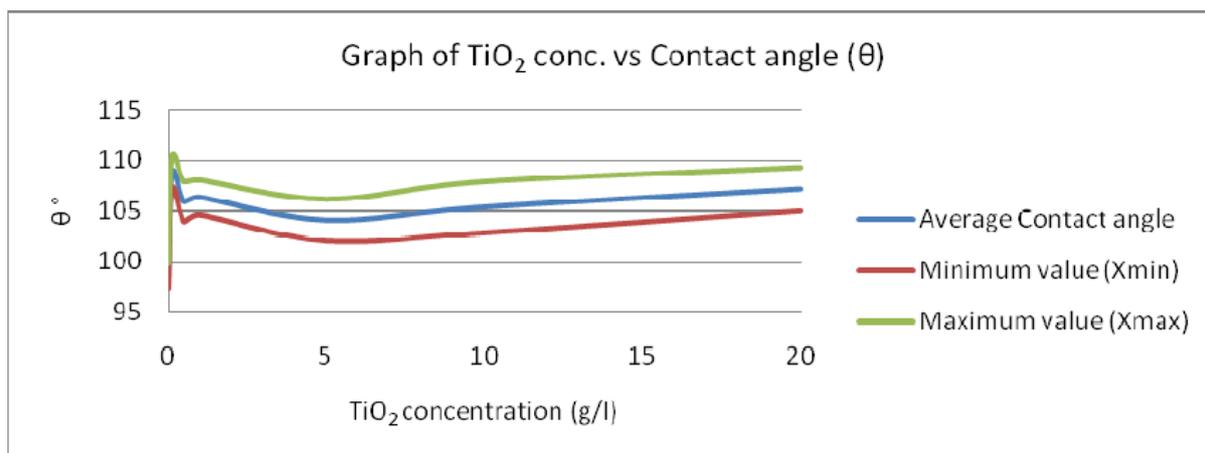
Hydrophobic Treatment of Glass Fibers with different TiO₂ powders

Figure 7. Description of the relation between the Contact angle and TiO₂ concentration for TiO₂-091009/1

In the results above the contact angle is high at 0.1g/l concentration and then decreases at 5g/l concentration. As it moves further to the increasing concentration the contact angle is constant.

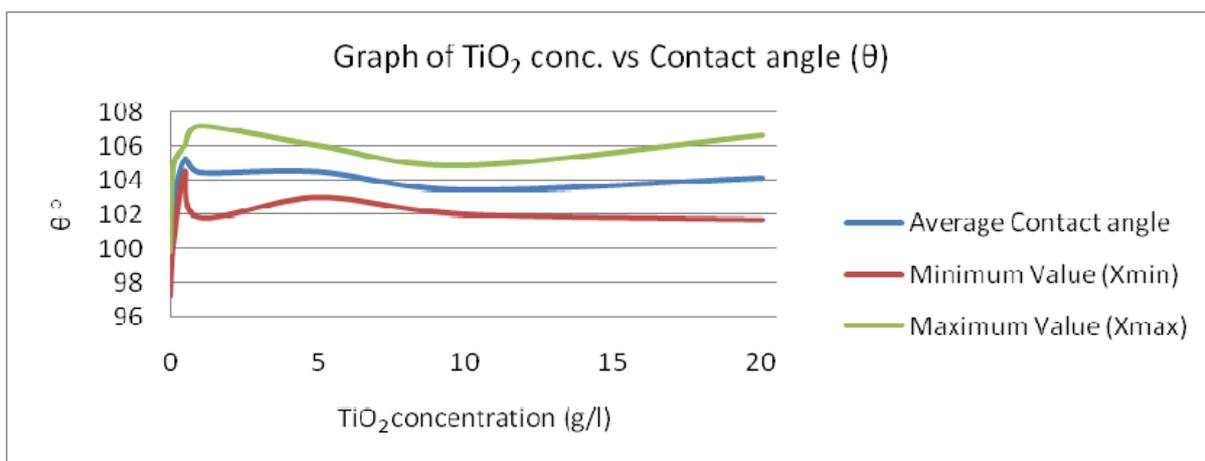


Figure 8. Description of the relation between the Contact angle and TiO₂ concentration for TiO₂-81109/2

The finishing treatment by the use of the TiO₂ -81109/2 the highest contact angle value at concentration 1g/l and the a decrease towards concentration 5g/l.and then a little bit of an increase but still less that at concentration 1g/l.

4. Conclusion

Super hydrophobic textile materials have been successfully prepared. The incorporation of TiO₂ particles by Titania sol finishing can not only cause a dual-size surface roughness for enhancing the hydrophobicity. All graphs for contact angle measurement show an increasing in contact angle from the lower concentration of TiO₂ to higher concentration.

5. References

1. Wiener J., Dembický J., *Wetting of anisotropic surfaces*, **Fibres and Textiles**,4, 2005, 167

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